

# TANDEM - Two Absorber Nanostructured DEvice Module for direct solar water splitting

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## INTRODUCTION

The sun offers the potential for harvesting enormous amounts of energy, although its present-day contribution to the energy mix is a miniscule 1-2% [1]. The amount of solar power striking the earth's surface at any one instant is equivalent to 130 million 500 MW power plants [2], nearly  $10^4$  times greater than our present day power needs. The ability to capture even a small percentage of this hugely abundant resource would render the current reliance on unnecessary CO<sub>2</sub> emitting fuel sources.

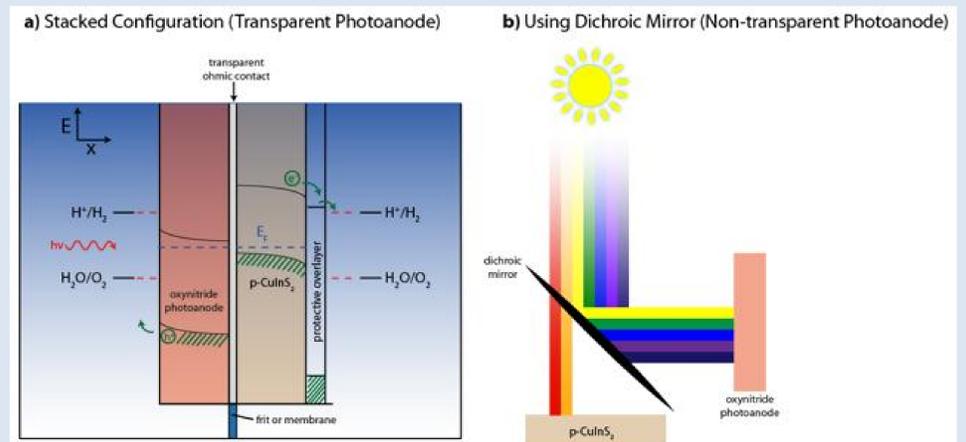
Photocatalytic or electrochemical water splitting, referred to very often as artificial photosynthesis, is a middle-to-long-term solution with tremendous promise. In analogy with nature's design, water is used as a feedstock of protons and electrons that are used to generate hydrogen fuel, a process which is driven by the energy contained in the solar photons. The energy-rich gaseous fuel product can participate directly in a hydrogen economy, or can be used to generate syn-gas, methane or ammonia, which is possible with present-day industrial scale technology. The strategy of generating a fuel, and not electricity, from sunlight alleviates the problems associated with electricity storage.

This project is designed to address major challenges that have thus far hindered the industrialization of artificial photosynthesis. It aims to find and improve cheap and efficient photocathode and photoanode materials for photoelectrochemical applications and to ensure long term operation with the implementation of corrosion resistant protective coatings.

## Goal: Unassisted solar water splitting by Tandem cells

The photoanode and the photocathode will be combined, together achieving the required photovoltage to enable unbiased overall solar water splitting.

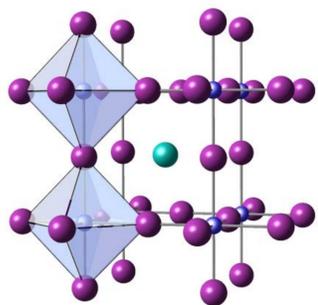
### Possible device architectures for Tandem cells



**Milestone after 6 months:** Photoanode: Fabrication of perovskite-based photocatalysts which absorb visible light for O<sub>2</sub> production. Photocathode: Fabrication of CuInS<sub>2</sub> photocathodes and exploratory studies with overlayer deposition by atomic layer deposition (ALD).

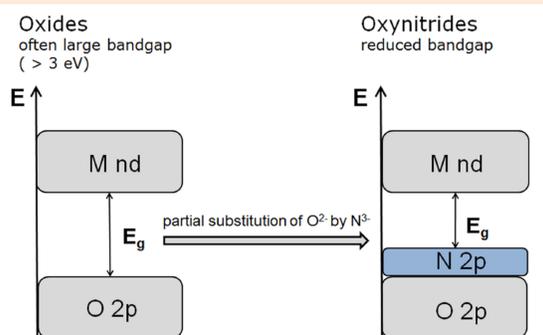
## Progress – 6 months

### Photoanode



Perovskite-type ABX<sub>3</sub> materials tolerate substitutions with a wide range of elements because of their flexible crystal structure.

e.g. SrTi<sub>1-x</sub>Nb<sub>x</sub>O<sub>2-y</sub>N<sub>y</sub> with 0.05 > x > 0.95 and 3.2 > E<sub>G</sub> > 1.8

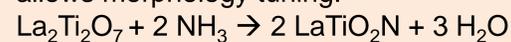


Anionic substitution and cross-substitution enable tuning of the band gap E<sub>G</sub>.



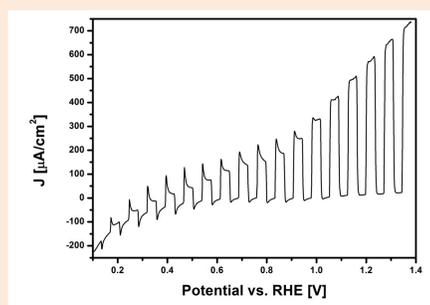
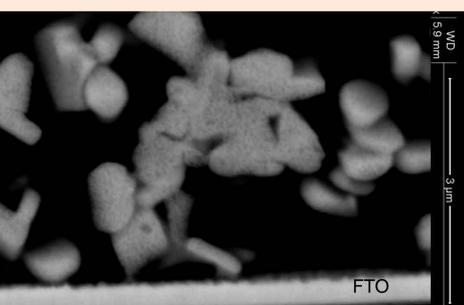
### Synthesis and Morphology tuning [4,5]

Soft chemistry and anionic substitution via thermal ammonolysis: A solid – gas reaction with NH<sub>3</sub> at T > 500 ° C which allows morphology tuning.



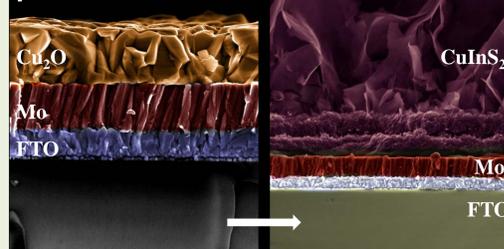
### Semi-transparent electrode fabrication with porous particles

### Photoelectrochemical performance of LaTiO<sub>2</sub>N electrode

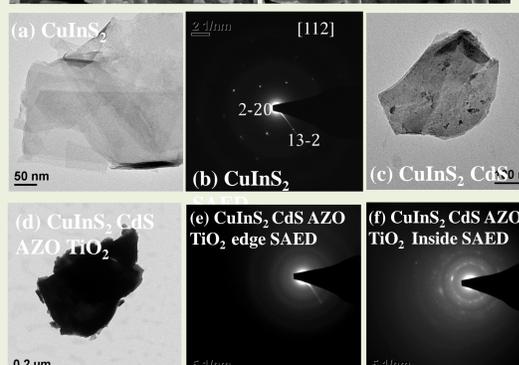
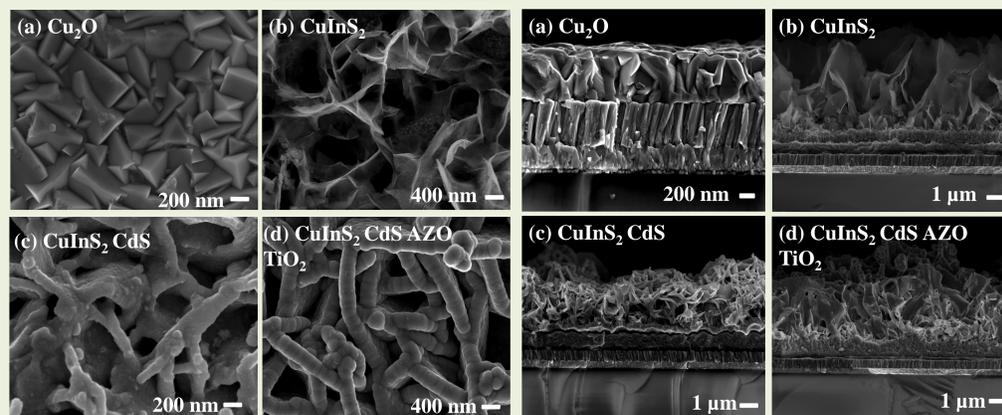
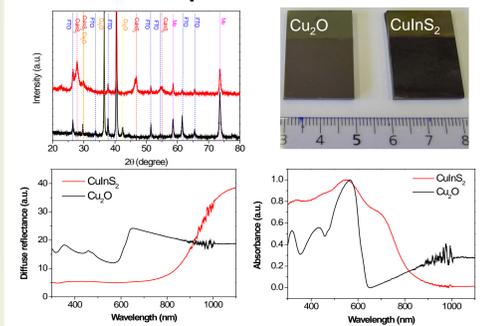


### Photocathode

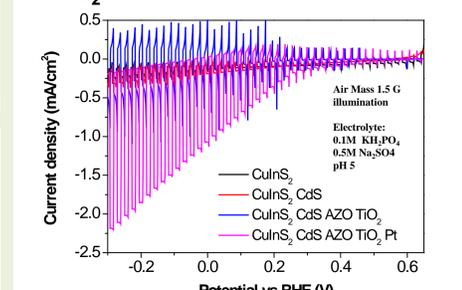
### Solution based transformation reaction route to fabricate CuInS<sub>2</sub> photocathode



### XRD and optical characterizations



### Photoelectrochemical performance of CuInS<sub>2</sub> electrode



## REFERENCES

[1] N. S. Lewis, G. Crabtree. (Office of Science, U. S. Department of Energy, Washington, DC, 2005).  
[2] M. G. Walter et al., *Chemical Reviews* **110**, (2010) 6446  
[3] A. Maegli, et al. *J. Solid State Chem.*, **184**, (2011) 929

[4] A. Maegli, et al., *Energy Procedia*, **22** (2012) 61  
[5] A.E. Maegli, et al. *J. Phys. Chem. C*, (2013) Ahead of Print.