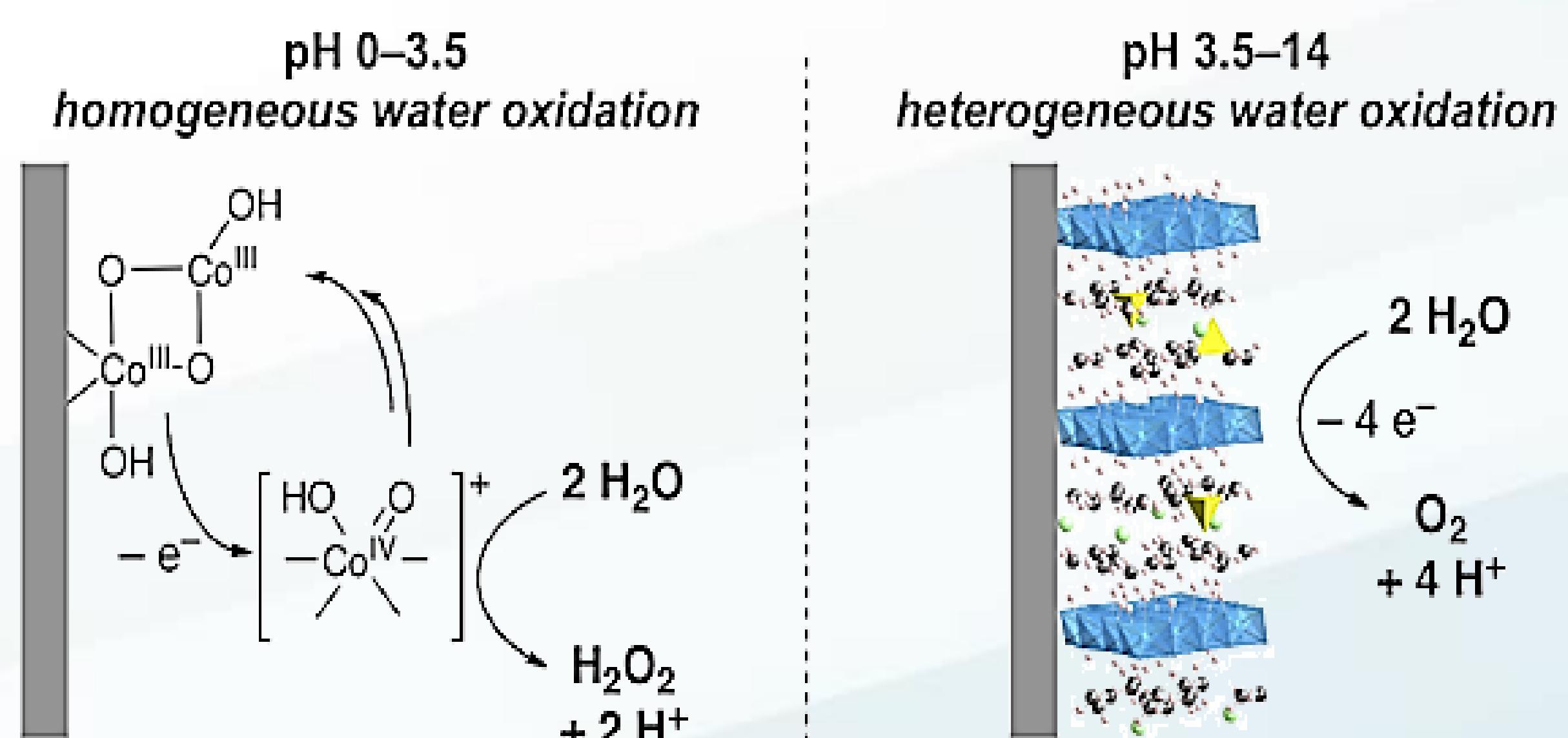


Electrocatalysts for the water oxidation reaction

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- Electrocatalysts facilitate redox reactions
- Oxygen evolution, hydrogen evolution and oxygen reduction reactions are relevant for renewable hydrogen economy
- *Design of active materials important aspect of electrocatalysis*
- Integrity and stability of electrocatalysts is a relevant issue
- We present materials and molecular based electrocatalysts

J. Am. Chem. Soc., 2011, 133 (36), pp 14431 – 14442

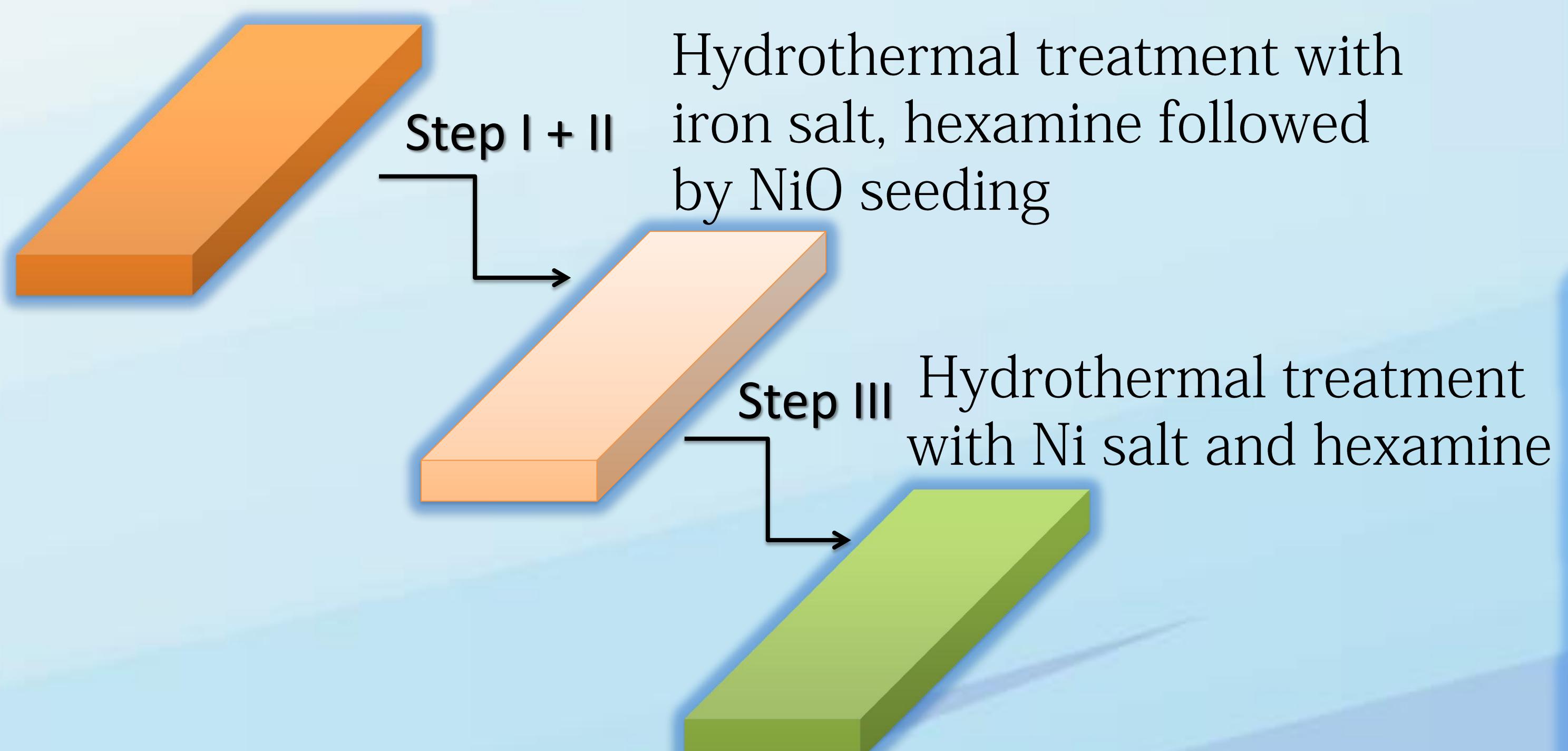


Fig. 1 : Synthesis of $\text{Ni}(\text{OH})_2$ coated electrode

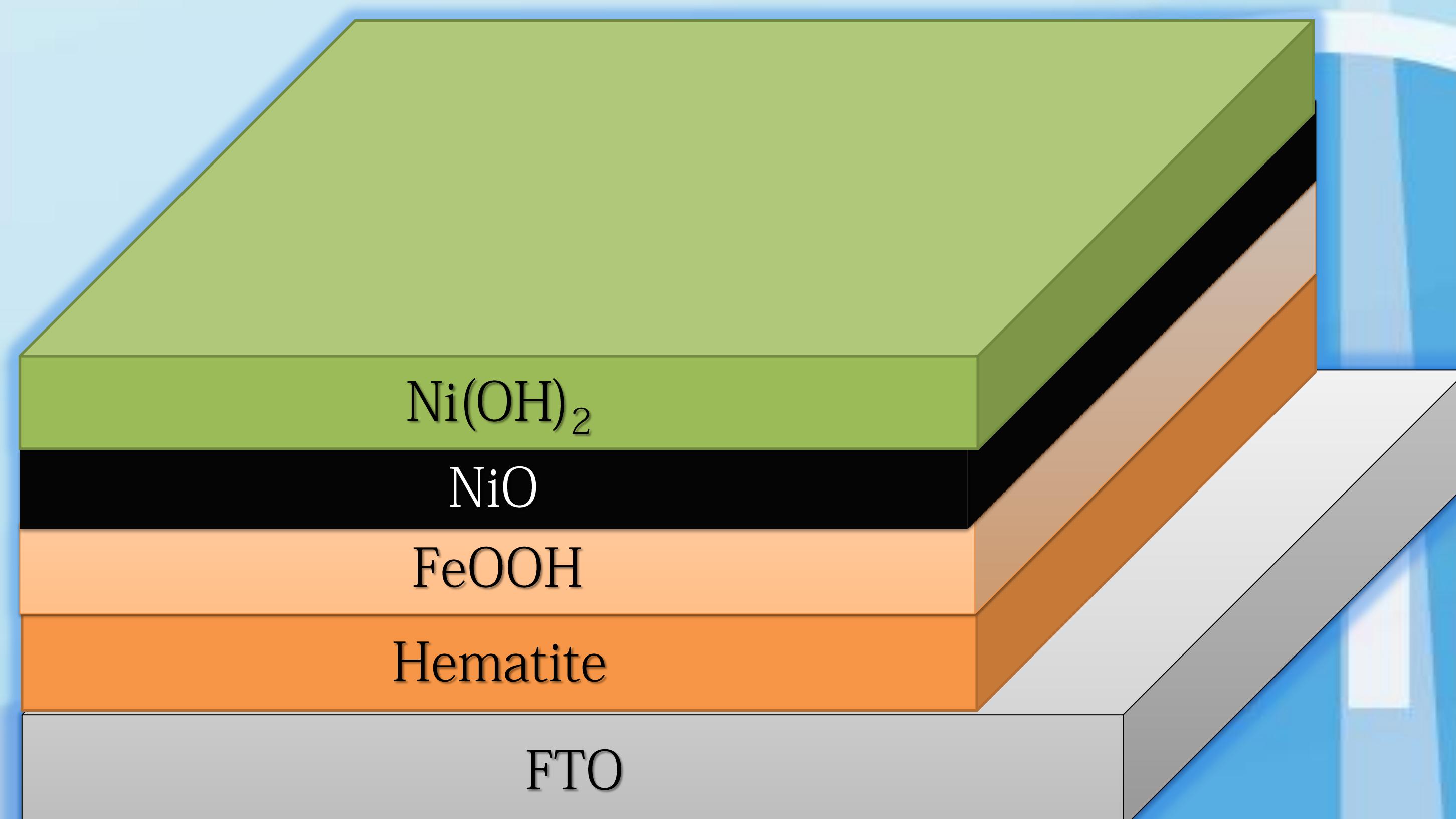


Fig. 2 : Sketch of a $\text{Ni}(\text{OH})_2$ electrocatalyst coated hematite electrode



Fig. 3: MEA development with $\text{Ni}(\text{OH})_2$ electrocatalysts



Fig. 4: SEM micrograph of $\text{Ni}(\text{OH})_2$ MEA
 MEA – membrane electrode assembly

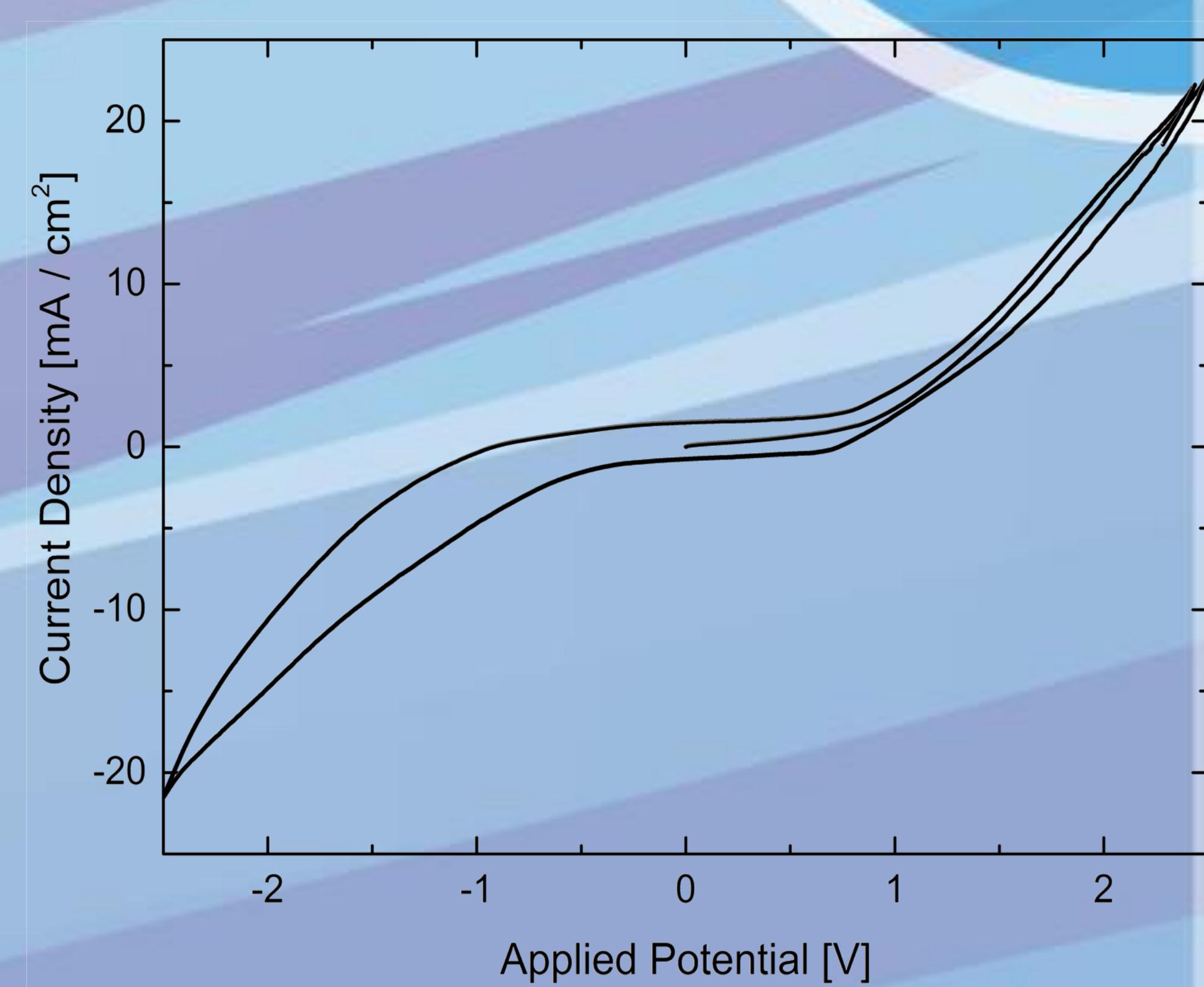


Fig. 5: cyclic voltammogram of $\text{Ni}(\text{OH})_2$ MEA

Self-assembled molecular water oxidation catalyst

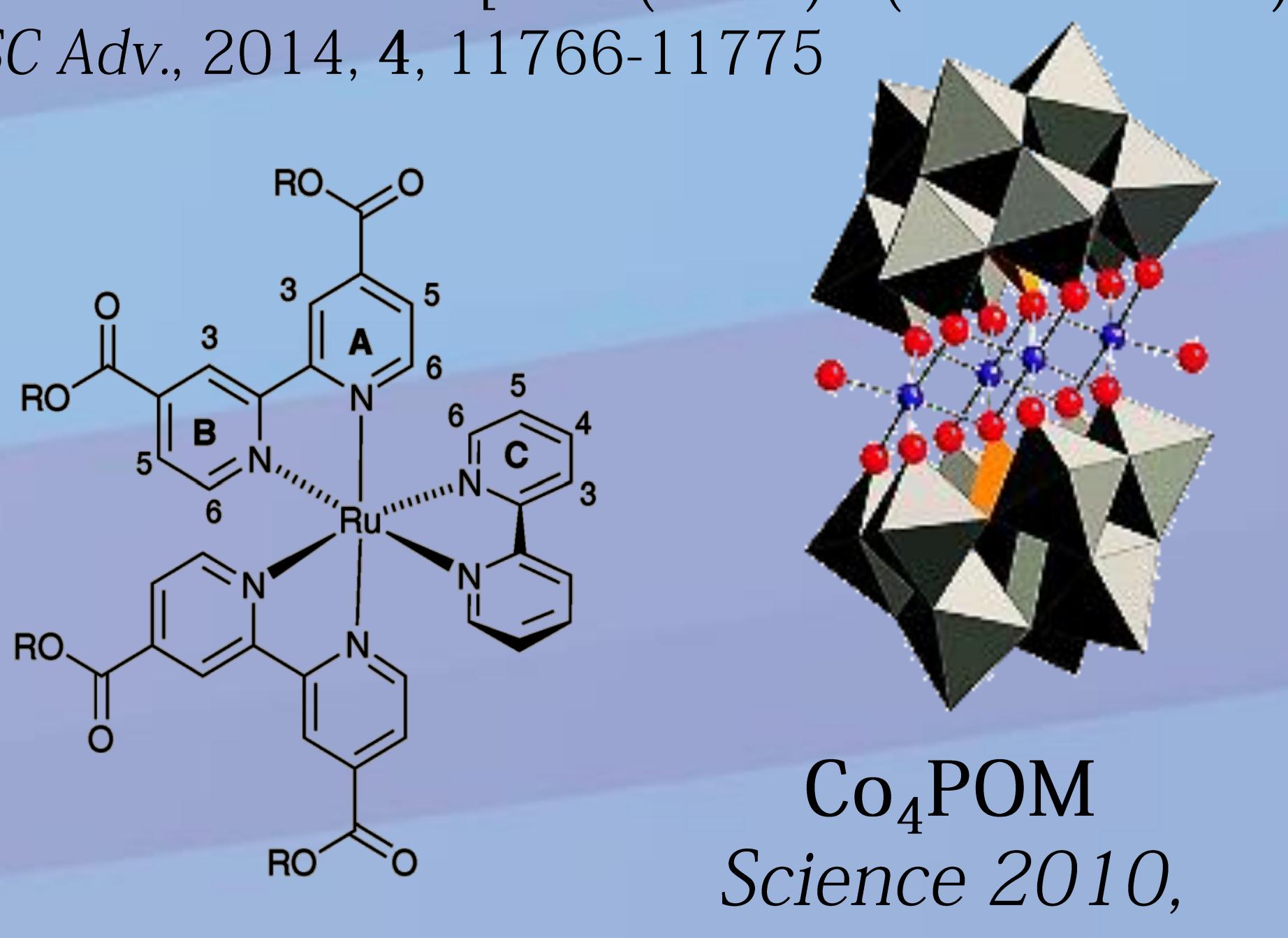
Langmuir-Blodgett (LB) films

of the WOC system:

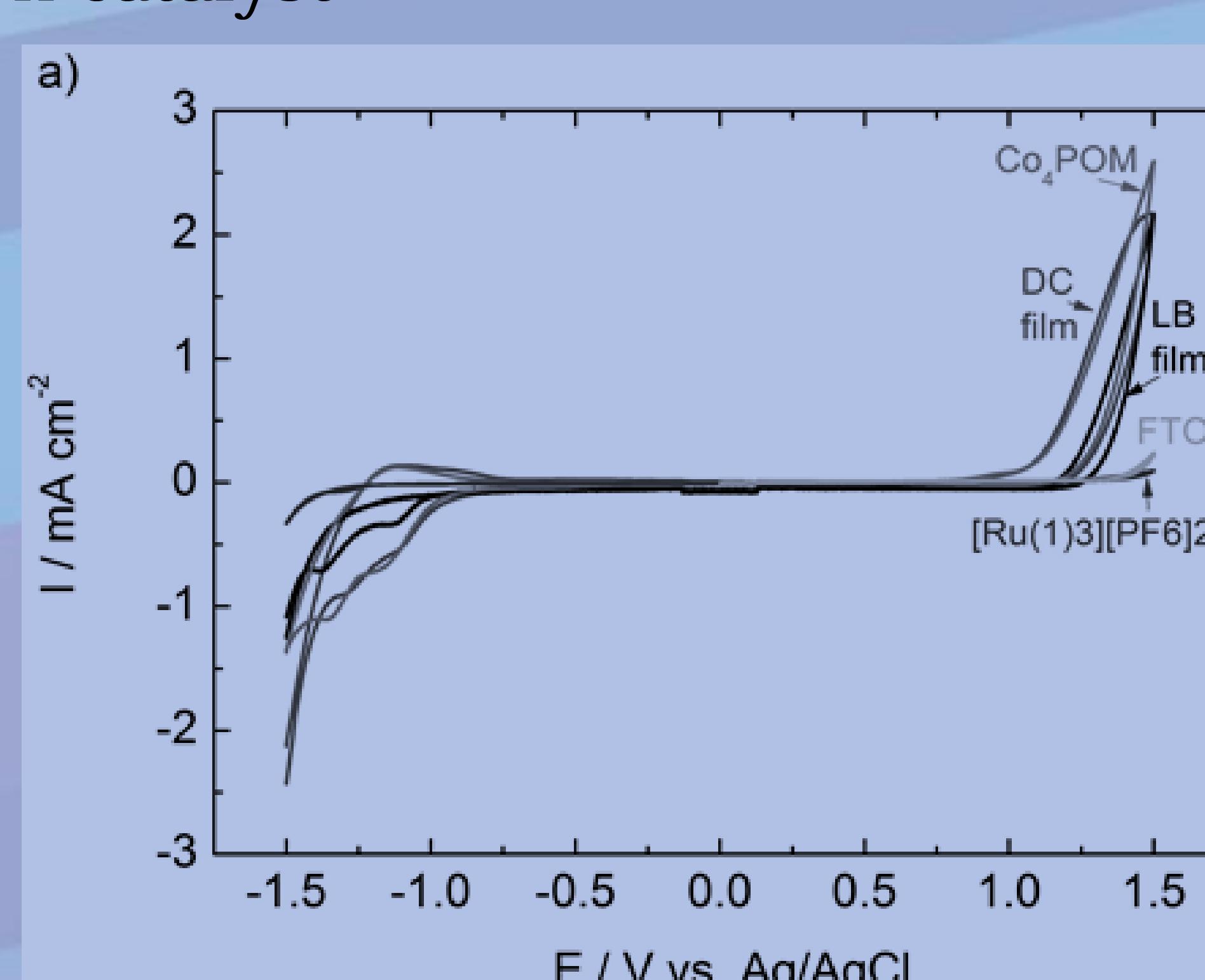
[Ru(1)₃][PF₆]₂ and Co₄POM

(Co₄POM = K₁₀[Co₄(H₂O)₂(α -PW₉O₃₄)₂])

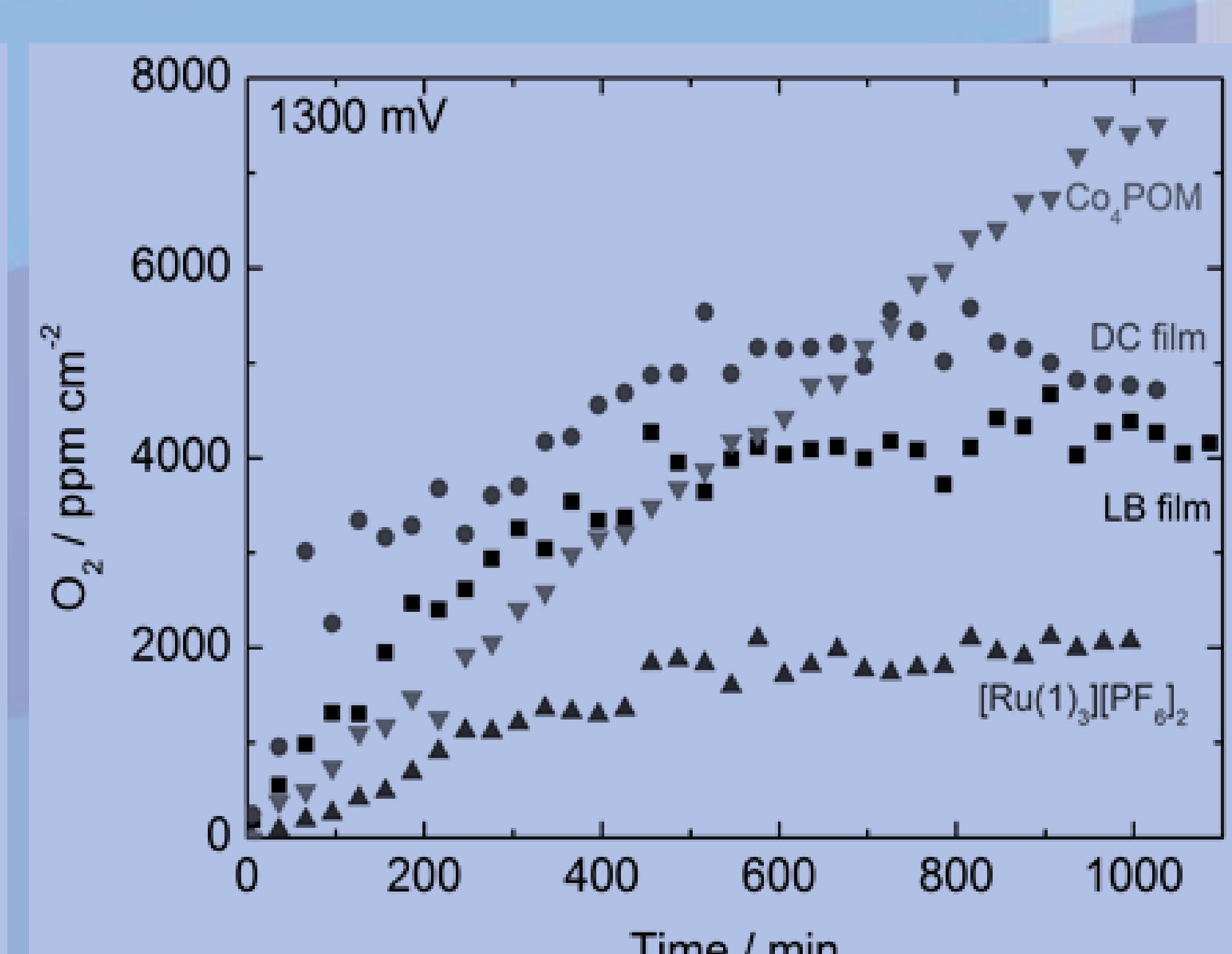
RSC Adv., 2014, 4, 11766-11775



Co₄POM
 Science 2010,
 328, 342-345



Cyclic voltammograms of 50-layer LB and drop cast films of WOC, [Ru(1)₃][PF₆]₂ and Co₄POM on FTO, and bare FTO



O₂ evolution vs time at +1.3 V bias for LB and drop cast WOC films, and separate Co₄POM and [Ru(1)₃][PF₆]₂ films